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09/870,037	05/30/2001	Hirokazu Yano	2204-010851	6743

7590

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EXAMINER

AHMED, SHEEBA

ART UNIT

PAPER NUMBER

1773

DATE MAILED: 03/20/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

A 2-6

# Office Action Summary

Application No.

09/870,037

Applicant(s)

YANO ET AL.

Examiner

Sheeba Ahmed

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-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

## Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

## Status

- 1) ☒ Responsive to communication(s) filed on 13 January 2003.
- 2a) ☒ This action is **FINAL**.                      2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

## Disposition of Claims

- 4) ☐ Claim(s) 1-19 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☐ Claim(s) 1-19 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

## Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
- Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
- 11) ☐ The proposed drawing correction filed on \_\_\_\_\_ is: a) ☐ approved b) ☐ disapproved by the Examiner.
- If approved, corrected drawings are required in reply to this Office action.
- 12) ☐ The oath or declaration is objected to by the Examiner.

## Priority under 35 U.S.C. §§ 119 and 120

- 13) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All   b) ☐ Some \* c) ☐ None of:
1. ☒ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).
- \* See the attached detailed Office action for a list of the certified copies not received.
- 14) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).
- a) ☐ The translation of the foreign language provisional application has been received.
- 15) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

## Attachment(s)

- 1) ☐ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO-1449) Paper No(s) \_\_\_\_\_.
- 4) ☐ Interview Summary (PTO-413) Paper No(s). \_\_\_\_\_.
- 5) ☐ Notice of Informal Patent Application (PTO-152)
- 6) ☐ Other: \_\_\_\_\_.

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## DETAILED ACTION

### *Response to Amendment*

1. Amendments to claims 1, 3, 8, and 16 have been entered in the above-identified application. **Claims 1-19 are pending.**

### *Claim Rejections - 35 USC § 112*

The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

2. Claims 3 and 16 are rejected under 35 U.S.C. 112, first paragraph, as containing subject matter which was not described in the specification in such a way as to reasonably convey to one skilled in the relevant art that the inventor(s), at the time the application was filed, had possession of the claimed invention.

Claims 3 and 16 have been amended to recite that the corrosion inhibitor is porous silica particles to which Ca ion is bonded at a ratio of 3-40 "weight %". The Applicants have directed the Examiner's attention to pages 3 and 4 of the Specification for support for such an amendment. However, pages 3 and 4 do not provide support for such an amendment. In fact, Page 4, lines 14-15 simply states that the "Ca ion is bonded to silica particles as the corrosion inhibitor A at a ratio of 3-40% in general". Furthermore, the Examiner was unable to find any other support in the original disclosure for the addition of the phrase "weight %" to claims 3 and 16. Applicants are required to cancel the new matter in the reply to this Office action and are reminded that

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the rejection of claims 3 and 16 under 35 USC 112, second paragraph may be re-instated upon cancellation of the new matter.

### ***Claim Rejections - 35 USC § 103***

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 1, 2, 4-10, 15, and 17-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Urata et al. (US 6,015,628) in view of Sasaki et al. (US 5,716,255).

Urata et al. disclose an organic composite coated steel sheet (***corresponding to the painted metal sheet of claim 8***) (Column 1, lines 8-10) comprising a rolled steel sheet base (***corresponding to the base metal sheet of claim 8***), a zinc or zinc-alloy plated layer (***corresponding to Zn coating of claim 9***), a chromate layer formed on the zinc or zinc alloy plated layer (***corresponding to the surface chemically treated with chromic acid as recited in claim 10***), and an organic film formed on the chromate layer. The organic film (***corresponding to the paint composition of the claimed invention***) comprises an epoxy resin (***corresponding to the base resin of the claimed invention and thus meeting the limitations of claims 2 and 15***) and a rust preventive additive (Column 9, lines 1-68 and Column 10, lines 46-63). The amount of the rust preventive additive is 3 to 50% by weight (***thus meeting the limitations of claims 4 and 17***) and examples include silica and aluminum dihydrogen

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tripolyphosphate (***corresponding to polyphosphate of the claimed invention and meeting the limitations of claims 5 and 18***) (Column 21, lines 20-40). Corrosion-inhibiting activity may be further improved by using ion-exchanged hydrophilic silica wherein the cation is calcium ion (***corresponding to Ca ion-exchanged porous silica claimed invention***) (Column 22, lines 23-35).

Urata et al. do not disclose that the organic film may comprise both the ion-exchanged silica and the aluminum dihydrogen tripolyphosphate.

However, Sasaki et al. disclose a sealing material which has better durability and comprises an aluminum phosphate and a silica wherein the amount of the silica is 15 to 50% by weight with the balance being the aluminum phosphate.

Accordingly, it would have been obvious to one having ordinary skill in the art to use both the ion-exchanged silica and the aluminum dihydrogen tripolyphosphate in the organic film disclosed by Urata et al. wherein the amount of the ion-exchanged silica is 15 to 50% by weight and the balance is aluminum dihydrogen tripolyphosphate given that Sasaki et al. teach that the combination of the silica and the aluminum phosphate in the claimed ratio provides better durability.

4. Claims 11-13 are rejected under 35 U.S.C. 103(a) as being unpatentable over Urata et al. (US 6,015,628) in view of Sasaki et al. (US 5,716,255) and Nagashima et al. (US 6,180,177 B1).

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Urata et al. and Sasaki et al., as discussed above, do not disclose that the base metal sheet has been treated with a fluoroacid such that fluorides are deposited on the surface of the base metal sheet as recited in claims 11-13.

However, Nagashima et al. disclose a surface treatment agents for metallic materials which can impart corrosion resistance and provides improved paint adherence to the metallic materials (Column 1, lines 10-15 and Column 8, lines 2-41). The surface treatment agent comprises fluoroacids which contain four or more fluorine atoms and one or more elements selected from the group consisting of titanium, zirconium, silicon, hafnium, aluminum and boron (Column 2, lines 50-60).

Accordingly, it would have been obvious to one having ordinary skill in the art to treat the base metal sheet disclosed by Urata et al. with a surface treatment agent comprising a fluoroacid which contains four or more fluorine atoms and one or more elements selected from the group consisting of titanium, zirconium, silicon, hafnium, aluminum and boron given that Nagashima et al. specifically teach that doing so can impart corrosion resistance and provides improved paint adherence to metallic materials. Furthermore, the Examiner takes the position that a surface treated with the surface treatment agent disclosed by Nagashima et al. would necessarily provide the deposited surface with fluorides in the claimed amount given that the chemical composition of the surface treatment agent and the manner of application as disclosed by Nagashima et al. and that of the claimed invention are identical.

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5. Claim 14 is rejected under 35 U.S.C. 103(a) as being unpatentable over Urata et al. (US 6,015,628) in view of Sasaki et al. (US 5,716,255) and Tanaka (US 5,623,003).

Urata et al. and Sasaki et al., as discussed above, do not disclose that the organic film layer may be coated with a topcoat.

However, Tanaka discloses a coating composition for metal substrates that provides corrosion resistance and comprises a polyester resin or an epoxy-modified polyester resin (Column 1, lines 10-15 and column 2, lines 12-15) and an anti-corrosion pigment such as aluminum dihydrogen tripolyphosphate (Column 6, lines 10-30) and which may be applied to a zinc-plated steel sheet (Column 6, lines 45-55) and may be further coated with a topcoating composition to further improve the corrosion resistance and appearance (Column 7, lines 1-7).

Accordingly, it would have been obvious to one having ordinary skill in the art to add a top coat to the organic film disclosed by Urata et al. given that Tanaka teaches that addend a topcoating layer further improves corrosion resistance and appearance.

### ***Response to Arguments***

6. Applicant's arguments filed on January 13, 2003 (Paper No. 5) have been fully considered but they are not persuasive. Applicants traverse the rejection of claims 1, 2, 4-10, 15, and 17-20 under 35 U.S.C. 103(a) as being unpatentable over Urata et al. (US 6,015,628) in view of Sasaki et al. (US 5,716,255) and submit that Urata teaches an organic film including rust preventive additives and only optionally utilizes silica ion-exchanged with calcium to improve the steel's corrosion-inhibiting activity and that Urata

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does not teach Ca-bonded silica particles dispersed in the base resin of the paint composition as recited by the now amended claims 1 and 8. However, the Examiner disagrees. Urata et al. specifically teach in Column 9, lines 1-68 and Column 10, lines 46-63 that the organic film (***corresponding to the paint composition of the claimed invention***) comprises an epoxy resin (***corresponding to the base resin of the claimed invention***) and a rust preventive additive wherein corrosion-inhibiting activity and high clearness of the final coated sheet can be obtained by using ion-exchanged hydrophilic silica wherein the cation is calcium ion (***corresponding to Ca ion-exchanged porous silica claimed invention***) as stated in Column 22, lines 23-35. Hence, silica ion-exchanged with calcium is a preferred corrosion-inhibiting agent and the base resin of the paint composition does in fact contain the rust preventive additive as recited by the now amended claims 1 and 8.

Applicants assert that Urata teaches away from the combination of the Ca-bonded silica particles and the aluminum dihydrogen triphosphosphate because Column 21, lines 38-40 state that the most preferred rust preventive additive is silica and chromate. Again, the Examiner disagrees with the Applicants interpretation of the Urata reference. The Examiner takes the position that Urata ***does not teach away*** from the combination of the Ca-bonded silica particles and the aluminum dihydrogen triphosphosphate. First, the proposed modification or combination does not change the principle of operation of Urata's invention and therefore the teachings of Urata and Sasaki are sufficient to render the claims *prima facie* obvious. Second, Urata falls far short of the kind of teaching that would discourage one of skill in the art from combining



the Ca-bonded silica particles with the aluminum dihydrogen triphosphate particularly given that Urata discloses both the Ca-bonded silica particles and the aluminum dihydrogen triphosphate as preferred rust preventive additives.

Furthermore, Applicants further assert that the Sasaki patent is non-analogous art since it is directed to sealing material. In response to applicant's argument that Sasaki is nonanalogous art, it has been held that a prior art reference must either be in the field of applicant's endeavor or, if not, then be reasonably pertinent to the particular problem with which the applicant was concerned, in order to be relied upon as a basis for rejection of the claimed invention. See *In re Oetiker*, 977 F.2d 1443, 24 USPQ2d 1443 (Fed. Cir. 1992). In this case, Sasaki et al. is directly pertinent to the particular problem with which the Applicants and Urata is concerned, i.e., corrosion inhibition.

Applicants further allege that the combination of Ca-bonded silica particles and polyphosphate in the base resin show unexpected results as provided in the Specification (particularly pages 3 and 4). The Examiner has reviewed the experimental data in the Specification and submits that the data is unclear, i.e., it is not readily apparent what ratios of Ca-bonded silica to polyphosphate result in improved properties. Furthermore, it appears that the claimed invention is not commensurate in scope with the experimental data provided, i.e., the ratio of Ca-bonded silica to polyphosphate is not claimed in independent claim 1 and 8. The Examiner requests that the Applicants specifically point out which data points show that the combination of

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Ca-bonded silica particles and polyphosphate in the base resin provides unexpected results and where such data is provided in the Specification.

### ***Conclusion***

7. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire **THREE MONTHS** from the mailing date of this action. In the event a first reply is filed within **TWO MONTHS** of the mailing date of this final action and the advisory action is not mailed until after the end of the **THREE-MONTH** shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than **SIX MONTHS** from the date of this final action.

8. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Sheeba Ahmed whose telephone number is (703)305-0594. The examiner can normally be reached on Mondays and Thursdays from 8am to 6pm.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Paul Thibodeau can be reached on (703)308-2367. The fax phone numbers for the organization where this application or proceeding is assigned are (703)305-5408 for regular communications and (703)305-3599 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is (703)306-5665.



Sheeba Ahmed  
March 17, 2003



Paul Thibodeau  
Supervisory Patent Examiner  
Technology Center 1700